

## **fraction extracted, $E$**

The fraction of the total quantity of a substance extracted (usually by the solvent) under specified conditions, i.e.  $E_A = \frac{Q_A}{Q'_A}$  where  $Q_A$  is the mass of **A** extracted and  $Q'_A$  is the total mass of **A** present at the start.

Notes:

1.  $E$  may be expressed as a percentage, %  $E$ .
2. The term extractability is qualitative and should not be used as a synonym for fraction extracted.
3. If the aqueous phase is extracted with  $n$  successive portions of solvent, the phase volume ratio (solvent/feed) being  $r$  each time, the fraction extracted is given by:

$$E_n = 1 - (rD + 1)^{-n}$$

If  $n = r = 1$  and  $E_1 = \frac{D}{1+D}$  this expression is a concept of value in chromatography theory.

4. The fraction extracted is also known as the recovery factor, especially for a multistage process.

### **Source:**

PAC, 1993, 65, 2373 (*Nomenclature for liquid-liquid distribution (solvent extraction)* (IUPAC Recommendations 1993)) on page 2384