

## endergonic (or endoergic) reaction

This is usually taken to be a reaction for which the overall standard Gibbs energy change  $\Delta G^\circ$  is positive. Some workers use this term with respect to a positive value of  $\Delta H^\circ$  at the absolute zero of temperature.

**Source:**

PAC, 1996, 68, 149 (*A glossary of terms used in chemical kinetics, including reaction dynamics (IUPAC Recommendations 1996)*) on page 164